



# Purification, Biochemical Characterization and Applications of *Pleurotus ostreatus* ARC280 Laccase

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## Authors' contributions

This work was carried out in collaboration between all authors. All authors contributed, read and approved the final manuscript.

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## ABSTRACT

**Aims:** To purify, characterize, and apply the laccase produced by submerged fermentation using an edible mushroom *Pleurotus ostreatus* ARC280.

**Study Design:** Laccase purification and characterization were designed using the most recent approaches and statistical studies of triplicate results values.

**Place and Duration of Study:** Department of Microbial Chemistry, Genetic Engineering and Biotechnology Division, National Research Centre, Dokki, Cairo, Egypt, between May 2011 and January 2013.

**Methodology:** *P. ostreatus* ARC280 laccase was purified using ammonium sulfate precipitation (40-80%), followed by gel filtration using Sephadex G100 column chromatography. The resulted pure laccase was analyzed on SDS-PAGE (12%). Laccase activity parameters such as temperature, pH, stability, metal ions and kinetic constants were studied. Laccase was applied to reduce four tumor cell lines growth and as antibacterial and antifungal agent.

**Results:** *P. ostreatus* ARC280 laccase was purified using ammonium sulphate followed by Sephadex G-100 chromatographic column by about 148 purification fold with Mr of 85kDa. Optimum *P. ostreatus* ARC280 purified laccase activity was recorded at 50°C and at pH 6.0, 3.0, 4.5 for Syringaldazine (SGZ), 2,2'-azinobis-(3-ethylbenzthiazoline-6-sulfonic) acid (ABTS) and 2, 6-dimethoxyphenol (DMP) as substrates, respectively. The purified enzyme

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was more stable in alkaline pH range and retained about 37.42, 73.51, 85.65, 87.7, 88.49, 93.65, 92.86 and 100.0 % of the initial activity after 5hrs of incubation at pH 3.0, 4.0, 5.0, 6.0, 7.0, 8.0, 9.0 and 10.0, respectively.  $\text{Hg}^{2+}$  caused complete inhibition at all tested concentrations; however  $\text{Mn}^{2+}$  ( $2.5 \times 10^{-3} \text{M}$ ) caused laccase activation by about 190 and 330% after 1 and 24 hrs, respectively.  $K_m$  and  $V_{max}$  were calculated and found to be 0.074, 2.857 and 0.476  $\mu\text{M}$  and 1.563, 2.500 and 2.632  $\mu\text{mol min}^{-1}$  for SGZ, DMP and ABTS, respectively. The purified enzyme has the ability to reduce four tested cell lines growth in vitro with percentage reduction of 16.8, 23.4, 15.2 and 23.4% for HePG2, HCT116, A549 and MCF7, respectively. On the other hand, the enzyme was found to have antibacterial and antifungal activities against *Escherichia coli* and *Candida albicans* respectively.

**Conclusion:** This enzyme seems to be a prospective enzyme for further biotechnological exploitation such as anticancer and antimicrobial activity applications.

*Keywords: Applications; characterization; laccase; pleurotus ostreatus; purification.*

## 1. INTRODUCTION

Laccases (E.C. 1.10.3.2) are N-glycosylated multi copper oxidases belonging to the group of the blue copper proteins [1,2]. They catalyze the oxidation of various substrates with concomitant reduction of molecular oxygen to water [3,4]. Laccase activity has been found in plants, insects, bacteria and fungi [4]. Different groups of fungi have been reported as ligninolytic enzymes producers [5]. However, most laccases are found and studied in lignin-degrading basidiomycetes [6,7]. Among the basidiomycetes, white-rot fungi have received great attention due to their powerful production of laccase, Mn-peroxidase and lignin-peroxidase [8,9].

Laccases are multinuclear enzymes [10] and the active site of laccase contains four copper atoms which are distributed in three sites, referred to as  $T_1$ ,  $T_2$  and  $T_3$  [11]. These copper atoms differ from each other in their paramagnetic resonance signals [12]. The laccase enzyme accepts electrons from substrates and converts them to free radicals. After receiving four electrons, the enzyme donates them to molecular oxygen to form two water molecules [13].

Recently, extensive attention has been focused on the application of laccases in different sectors [14]. The applications of laccases have arisen due to its broad substrate range, and the conversion of substrates to unstable free radicals that may undergo further non-enzymatic reactions [15]. Laccases are found to be used in dyes decolorization, bioremediation and biodegradation, paper and pulp industry, food processing industry, deodorants, toothpastes, mouthwashes and detergents [16,17]. Applications of laccases in the synthesis of complex medicinal compounds as well as heteromolecular dimers of antibiotics via phenolic oxidation [18], phenolic oxidative coupling [19] and oxidation coupled with nuclear amination were reported [20]. Laccases are also involved in enzyme-catalyzed production of anticancer drugs [21]. Antibody or antigen-conjugate laccase can be used as a marker enzyme for immunochemical assays [22]. The present study was aimed to purify and characterize a laccase from *P. ostreatus* ARC280 liquid cultures and to assess its possible antitumoral and antimicrobial activity.

## 2. MATERIALS AND METHODS

### 2.1 Microorganisms

*P. ostreatus* ARC280 was obtained from Agriculture Research Center (ARC), Egypt. A Gram positive bacterial strain *Bacillus mycooides*, a Gram negative bacterial strain *Escherichia coli*, and a non-filamentous fungus *Candida albicans* were selected to test the antimicrobial activity of laccase.

### 2.2 Media

*P. ostreatus* ARC280 was maintained and periodically sub-cultured on potato-dextrose agar (PDA). The liquid medium described by Tlecuil-Beristain et al. [23] was used for laccase production with using soluble starch instead of glucose as a carbon source. The pH value was adjusted to 5.0 before autoclaving, after that the medium was inoculated with *P. ostreatus* ARC280 and incubated statically at 28°C for 26 days, after that the filtrated medium was used as a source of enzyme for purification. *B. mycooides*, *E. coli* and *C. albicans* were cultivated and kept on slants of modified nutrient agar medium (g/l): peptone, 3.0; yeast extract, 1.5; meat extract, 1.5; glucose, 0.5; NaCl, 0.25 and agar, 20.0. The pH value was adjusted to 7.0 before autoclaving.

### 2.3 Chemicals

Laccase substrates were supplied by Sigma-Aldrich; USA: SGZ, Aldrich W404901; ABTS, Sigma A1888 and DMP, Sigma D135550. All chemicals used in this study were of analytical grade.

### 2.4 Laccase Assay

The enzyme activity was assayed by monitoring the rate of SGZ oxidation spectrophotometrically at 525 nm [24]. The reaction mixture (2.0 ml) contained SGZ, 0.1 µmole; citrate-phosphate buffer (pH 6.0), 90 µmoles and appropriate volume of diluted enzyme. Activity unit was defined as the change in the absorbance of 0.001/sec [7,25]. In the experiments of optimum pH and substrate specificity, the activity was determined using: 1 mM SGZ [4,24], 5mM ABTS [2, 26] and 5mM DMP [27,28]. The enzyme activity was calculated according to the method described by Annuar *et al.* [29] as follows:

$$\text{Laccase activity (Units/L)} = \frac{\Delta Abs}{\Delta t \ \epsilon \ l} \times \frac{\text{Total assay volume}}{\text{Enzyme sample volume}}$$

Where  $\Delta t$  is the time of incubation (min),  $\Delta Abs$  is the change in absorbance,  $\epsilon$  is the extinction coefficient of substrates ( $\epsilon_{525} = 65,000$  for SGZ;  $\epsilon_{436} = 36,000$  for ABTS and  $\epsilon_{469} = 49,600$  for DMP) in units of  $M^{-1} \text{ cm}^{-1}$ , and  $l$  is the cuvette diameter (1cm). Enzyme activity (Unit) was defined as the amount of enzyme that oxidized 1µmol of substrate/min. The oxidation rates were determined at room temperature (25°C±2) in triplicate. The protein in the crude enzyme preparations was determined by the modified procedure of Lowry et al. [30]. Determination of protein in the pure enzyme preparations was made according to Layne [31] using the following equation:

$$\text{Protein (mg/ml)} = 1.55 \times A_{280} - 0.76 \times A_{260}$$

All the data was statistically evaluated according to the method described by Kenney and Keeping [32], where the means and standard deviations (Mean  $\pm$  S.D.) were calculated for each experiment.

## 2.5 Purification of *P. ostreatus* ARC280 Laccase

Ammonium sulphate (40-80%) was used to precipitate the laccase protein produced by *P. ostreatus* ARC280 after 26 days of incubation. The precipitate was separated from the filtrate by centrifugation at 12000 rpm for 15 min in a pre-cooled rotor (HERMLE Z 323 K cooling centrifuge) at 0°C. The precipitated protein was dissolved in citrate-phosphate buffer (0.05M, pH 6.0), dialyzed and then applied to Sephadex G-100 column chromatography (3.0 $\times$ 26 cm) which had been equilibrated with 0.05 M citrate-phosphate buffer (pH 6.0). Five milliliters fractions were collected at a flow rate of 0.5ml/min.

## 2.6 Electrophoretic Analysis (SDS-PAGE)

Sodium dodecyl sulphate–polyacrylamide gel electrophoresis (SDS-PAGE) was used to monitor the development of the purification process and to determine the apparent molecular mass of the laccase enzyme. SDS-PAGE was performed with 12% polyacrylamide according to Laemmli [33]. Proteins were visualized by staining for 3 hours with Coomassie Brilliant Blue-R250. Gel was then destained with a mixture of acetic acid and ethanol (40%: 10%). Diluted protein samples were concentrated by the use of a lyophilizer (Snijders Scientific b.v., L45FM – RB, Tilburg - Holland). Low molecular weight markers; phosphorylase b (94 kDa), bovine serum albumin (67 kDa), ovalbumin (43 kDa), carbonic anhydrase (30 kDa) and soybean trypsin inhibitor (20.1 kDa) were used.

## 2.7 Effect of Temperature and Thermal Stability

The dependence of the activity of the purified enzyme on temperature had been assessed at different incubation temperatures (10, 20, 30, 40, 50, 60, 70 and 80°C) at pH 6.0. For thermal stability behavior, similar aliquots of the purified enzyme were first incubated separately at different temperatures (30, 40, 50, 60, 70 and 80°C) for different incubation periods. The residual enzyme activity was then determined by incubating each preheated enzyme aliquot with the remaining components of the standard reaction mixture, containing 0.1 $\mu$ mole of SGZ as the substrate.

## 2.8 Effect of pH Value and pH Stability

The effect of pH on the activity of the purified laccase of *P. ostreatus* ARC280 was examined using different pH values of 0.1M citrate-phosphate buffer (pH 2.6 – 7.0) with SGZ, DMP or ABTS individually as a substrate. In connection to pH stability, aliquots of the pure laccase were incubated with different buffer systems (0.1M) at different pH values ranging from pH 3.0 to 10.0 (citrate phosphate buffer (3.0–7.0); phosphate buffer (8.0) and bicarbonate buffer (9.0–10.0) for different incubation periods up to 28 h. Identical aliquots of enzyme were removed at different time intervals and assayed for residual laccase activity using SGZ as a substrate at pH 6.0.

## 2.9 Effect of EDTA and Different Metal salts

EDTA and different metal salts of various cations ( $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mn}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Cu}^{2+}$  and  $\text{Zn}^{2+}$ ) were added separately at a final concentrations of  $2.5 \times 10^{-3}$ ,  $5 \times 10^{-3}$ ,  $1 \times 10^{-2}$  and  $5 \times 10^{-2} \text{M}$  to each reaction mixture containing purified laccase. The purified enzyme was incubated with EDTA or metal salts and buffer at room temperature for different time intervals, and then SGZ was added as substrate to each reaction mixture and assayed for laccase activity using the standard assay method.

## 2.10 Substrate Specificity and Kinetic Constants

The activity of *P. ostreatus* ARC280 purified laccase was examined towards three different substrates (SGZ, DMP and ABTS) by incubating the purified enzyme with each individual substrate at concentrations of 1mM SGZ [24] at pH 6.0, 5mM DMP [28] at pH 4.5 and 5mM ABTS [26] at pH 3.0. The enzyme activity was assayed by the method described by Annuar et al. [29]. Kinetic constants such as  $K_m$  and  $V_{max}$  were determined using various concentrations of different substrates (SGZ, DMP and ABTS). The  $V_{max}$  values were calculated from the dependence of the activity on the concentration of the substrate and the  $K_m$  values were calculated from Lineweaver-Burk plot.

## 2.11 Cytotoxic Effect of Laccase on Human Tumor Cell Lines

Cell viability was assessed by the mitochondrial dependent reduction of yellow MTT (3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide) to purple formazan [34]. Cells were suspended in RPMI 1640 medium for HePG2 (Human hepatocellular carcinoma cell line), HCT116 (Colon cell line) and MCF7 (Human caucasian breast adenocarcinoma); and DMEM for A549 (Lung carcinoma cell line). The media were supplemented with 1% antibiotic-antimycotic mixture (10,000Units/ml potassium penicillin, 10,000 $\mu\text{g}/\text{ml}$  streptomycin sulfate and 25 $\mu\text{g}/\text{ml}$  amphotericin B), 1% L-glutamine and 10% fetal bovine serum and kept at 37°C under 5%  $\text{CO}_2$ . Cells were batch-cultured for 10 days, then seeded at a concentration of  $1 \times 10^3$  cells/well in a fresh growth medium in microtiter plastic plates at 37°C for 24h under 5%  $\text{CO}_2$  [35]. Media was aspirated and the cells were incubated either alone (negative control) or with different concentrations of sample. After 48h of incubation, medium was aspirated, 40  $\mu\text{l}$  MTT salt (2.5 $\mu\text{g}/\text{ml}$ ) were added to each well and incubated for further four hours at 37°C under 5%  $\text{CO}_2$ . Two hundred  $\mu\text{l}$  of 10% sodium dodecyl sulphate (SDS) in deionized water was added to each well and incubated overnight at 37°C, to stop the reaction and dissolving the formed crystals. A positive control (Adrinamycin (Doxorubicin)) [Mw= 579.99] which composed of 100 $\mu\text{g}/\text{ml}$  was used as a known cytotoxic natural agent who gives 100% lethality under the same conditions [34]. The absorbance was then measured using a microplate multi-well reader (Bio-Rad Laboratories Inc., model 3350) at 595 nm and a reference wavelength of 620nm. The percentage of change in viability was calculated according to the formula:

$$((\text{Reading of extract} / \text{Reading of negative control}) - 1) \times 100$$

## 2.12 Antimicrobial Activities

*B. mycoides*, *E. coli*, and *C. albicans* were used as model microorganisms. Cultures were prepared by inoculation of nutrient agar medium with 100 $\mu\text{l}$  of re-suspended overnight culture at 37°C ( $1 \times 10^7$  CFU/100 $\mu\text{l}$ ). The nutrient agar medium was poured in Petri dishes,

solidified, then wells in solidified agar plates were made and equal units of the tested laccase preparations (1.5 Units/ml) were inserted in each well, then plates were incubated at 37°C for 7h [36].

### 3. RESULTS AND DISCUSSION

#### 3.1 Purification of *P. ostreatus* ARC280 Laccase

*P. ostreatus* ARC280 crude laccase with 464.67U/ml, as we described in our previous work [2], underwent fractional precipitation with ammonium sulphate within 40-80% saturation with a purification fold of 6.34 and a recovery of 80.63% (Table 1). The laccase obtained from ammonium sulphate precipitation was applied to Sephadex G-100 chromatographic column. Laccase from *P. ostreatus* ARC280 was successfully well purified using Sephadex G-100 column (Fig. 1). Optimum purification fold (147.74) was obtained with fraction F<sub>13</sub> with a recovery percentage of 11.82% (Table 1).

**Table 1. Purification of *P. ostreatus* ARC280 laccase**

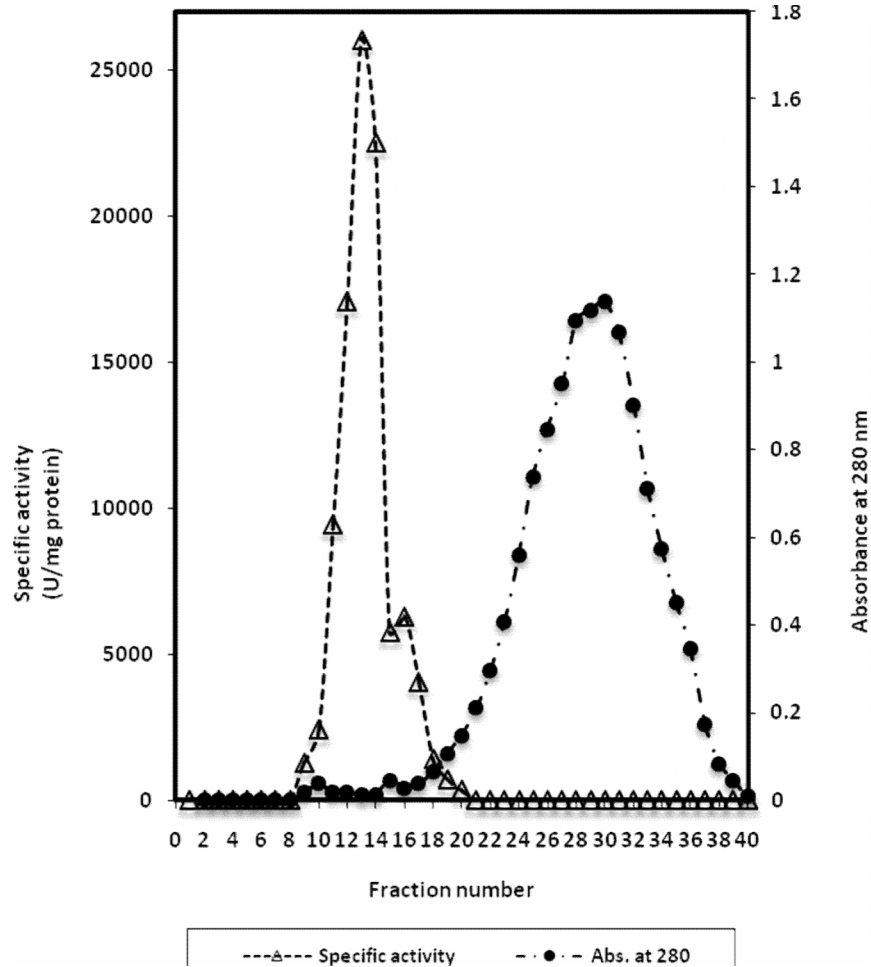
Fraction	Total protein (mg)	Laccase activity			
		Total units	Specific activity (Units/mg protein)	Recovery (%)	Purification fold
Crude	137.50	24250.00	176.36	100.00	1.00
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> (40-80%)	17.50	19553.33	1117.33	80.63	6.34
<b>Sephadex G-100 column</b>					
F <sub>12</sub>	0.17	2906.57	17097.47	11.99	96.95
F <sub>13</sub>	0.11	2866.14	26055.82	11.82	147.74
F <sub>14</sub>	0.11	2476.64	22514.19	10.21	127.66

In this connection, Patel et al. [37] purified the laccase produced from *P. ostreatus* HP-1 with DEAE-sepharose column to obtain 13.13-fold purification with 77.63% yield of laccase enzyme. Palmieri et al. [38] achieved 85-fold purification from *P. ostreatus* POXA1 laccase isoenzyme with a final yield of 23%, and a lower yield for POXA2 isoenzyme. Murugesan et al. [39] purified laccase from *P. sajor-caju* using ammonium sulfate (70% w/v), DEAE-cellulose, and Sephadex G-100 column chromatography with an overall yield of 53% and a 10.3 purification fold. Adamafio et al. [40] purified laccase from *P. ostreatus* strain EM-1 using ammonium sulphate precipitation and gel filtration using Sephadex G-75 with a purification fold of 12.7 and a recovery yield of 21%.

#### 3.2 SDS-PAGE Analysis

SDS-PAGE analysis indicated that the protein isolated from *P. ostreatus* ARC280 is a monomer in solution with a molecular mass of 85kDa (data not shown) which is similar to other fungal laccases. The molecular masses of *P. ostreatus* strain V-184 laccases LCC3 and LCC4 were around 80 and 82kDa, respectively [41]. In this connection, many investigators [42,1,43] reported the purification of laccase from *Scytalidium thermophilum*, *Paraconiothyrium variable* and *Trichoderma harzianum* with molecular masses of 82, 84 and 79kDa, respectively. In contrast, Asgher et al. [44] and Xu et al. [45] purified laccases from *Trametes versicolor* and *Lentinus tigrinus* with relative molecular masses of 63 and 59 kDa, respectively. In this connection, *P. ostreatus* strain V-184 laccases LCC1 and LCC2 have molecular masses of about 60 and 65 kDa, respectively, [41]. The purified laccase from

*P. ostreatus* HP-1 and *P. ostreatus* D1 were shown to be with relative molecular masses of 68kDa and 64kDa, respectively [37,46].



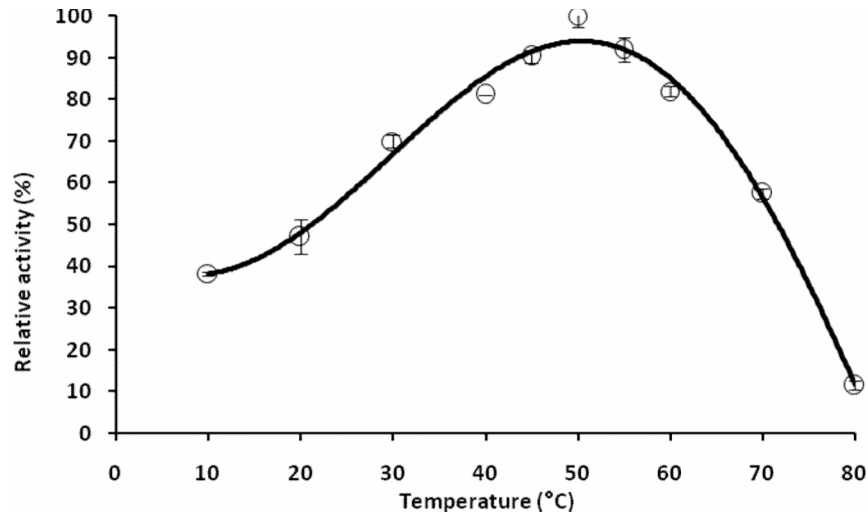
**Fig. 1. Gel filtration of ammonium sulphate partially purified laccase fraction (40-80%) of *P. ostreatus* ARC280 using G-100 Sephadex column**

### 3.3 Effect of Temperature on Laccase Activity and Stability

Tinoco et al. [47], stated that the temperature profile for all *P. ostreatus* laccases showed maximal activity between 30 and 40°C, and the activity was drastically reduced when the reaction temperature was raised to 60°C. In the present study, *P. ostreatus* ARC280 purified laccase displayed a maximum activity at 50°C (Fig. 2), which is similar to the optimum temperature determined by Forootanfar et al. [1] for laccase from *Paraconiothyrium variable*, *P. ostreatus* HP-1, *P. ostreatus* strain EM-1 and *P. ostreatus* [37,40,48]. On the other hand, optimum activity of the purified laccase from *Hericium coralloides* and *Trametes versicolor* IBL-04 could be detected at 40°C [44,49], whereas the optimum activity for laccase from *Trametes hirsuta* Bm-2 is ranged from 40 to 60°C [50] and the laccase from *P. ostreatus* HP-1 was active in the temperature range of 40–70°C [37]. In other approaches, maximal

laccase activity was observed at temperatures of 60 and 80°C for laccase from the mushroom *Lentinus tigrinus* [45] and LacIII isozyme from the white rot fungus *Trametes* sp. HS-03 [51], respectively. Palmieri et al. [38] found that the *P. ostreatus* laccase isoenzyme POXA1 showed maximal activity in the range 45–65°C, whereas POXC showed maximal activity in a narrower range (50 – 60°C) and POXA2, at a lower temperature (25–35°C).

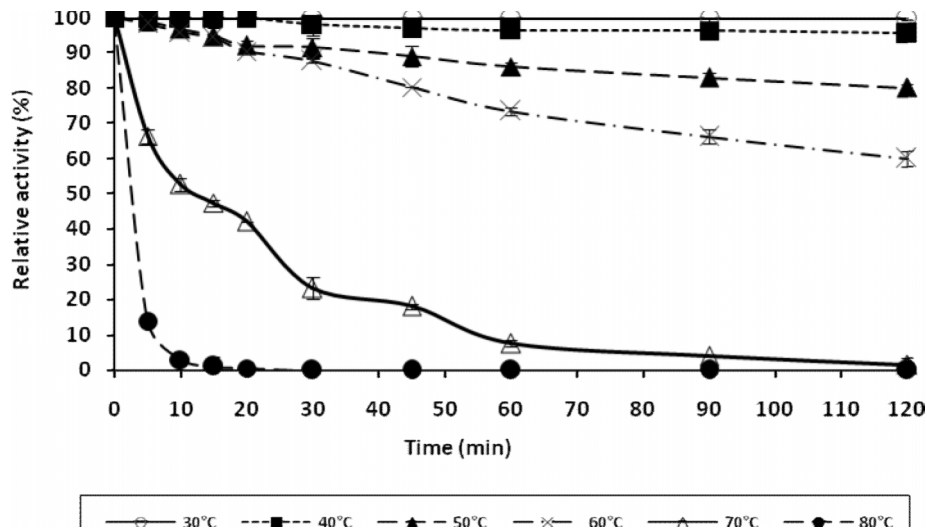
Depending on the source of the enzyme, thermal stability varied at different temperatures. In general, typical fungal laccases exhibited a half-life of 1 h at 70°C and below 10 min at 80°C [43]. The present studies revealed that the *P. ostreatus* ARC280 purified laccase could sustain heating up to 30 without apparent loss of activity for 120 min, and the residual activity was regularly decreased as a function of both time of exposure and temperature. Incubating the enzyme at 40, 50 and 60°C for 120min resulted in a loss of about 4, 20 and 40% of its activity, respectively (Fig. 3). These results indicated that the enzyme under investigation is more stable than *P. ostreatus* strain EM-1 [40] and *Paraconiothyrium variable* [1] laccases as they retained only 22.6 and 50% of their initial activity after 60 min of incubation at 50°C respectively, and it resembles laccase isolated from *Streptomyces cyaneus* which is reported to have retained more than 75% of its activity after incubation for 120min at 50°C [52]. Luna et al. [53], stated that the laccase from *P. ostreatus* showed greater stability at 39°C and pH 6, with value of residual activity of 27.5% after 15 hours of incubation. Laccase from *P. ostreatus* HP-1 was found stable for 10min at 50°C and considerable loss of activity was observed after 12 min. In the initial 15 min of incubation, the half of the laccase activity was found to be lost at 60°C, whereas rapid inactivation of the enzyme was observed at 70°C [37].



**Fig. 2. Effect of reaction temperature on the activity of *P. ostreatus* ARC280 purified laccase**

Reaction mixture contained: SGZ, 0.1  $\mu$ mole; purified enzyme, 1.0 $\mu$ g protein; citrate-phosphate buffer pH 6.0, 90  $\mu$ moles; temperature, as indicated; total volume, 2ml. Values are means of three replications  $\pm$  standard deviation

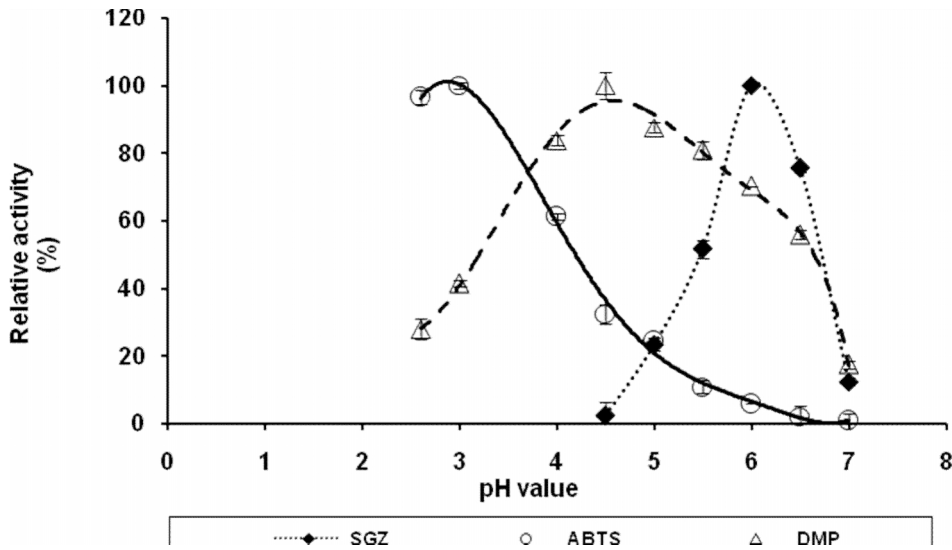




**Fig. 3. Thermal stability behavior of the *P. ostreatus* ARC280 purified laccase**  
 Reaction mixture contained: SGZ, 0.1  $\mu$ mole; purified enzyme, 3.0  $\mu$ g protein; citrate-phosphate buffer pH 6.0, 90  $\mu$ moles; time of exposure of the enzyme at different degrees of temperature, as indicated; total volume, 2ml. Values are means of three replications  $\pm$  standard deviation

### 3.4 Effect of pH Value on Laccase Activity and Stability

The pH optima for laccases can vary depending on the substrate used and on its redox potential [54]. Fungal laccases are usually stable at acidic pH, although pH stability varies considerably depending on the source of the enzyme [55]. In the present study, the effect of pH on the activity of the purified laccase of *P. ostreatus* ARC280 was examined using SGZ, DMP or ABTS as substrates. The obtained results showed that, the optimum activity was recorded at pH 6.0, 4.5 and 3.0 for SGZ, DMP and ABTS, respectively (Fig. 4). In agreement with these results, Schücker et al. [56] reported that, the highest laccase activity for the oxidation of ABTS, SGZ, guaiacol and DMP was found at pH 3.0, 6.0, 5.5 and 4.0, respectively. The pH profile of *P. ostreatus* HP-1 laccase showed maximum activity at pH 4.5 when ABTS was used as a substrate [37]. The results on pH optima are consistent with previous reports that suggest that this value depends mainly on the substrates used. The optimal pH range for fungal laccases activities on substrates like DMP, guaiacol and syringaldazine, were found to be between 4.0 and 7.0 [55,57]. Patrick et al. [58] stated that, the pH profile is the result of two opposing effects: The first effect is due to the redox potential difference between a reducing substrate (phenolic compound) and the Type 1 copper center of laccase, where the electron transfer rate is favored for phenolic substrates at a high pH. The second effect is generated by the binding of a hydroxide anion to the type 2/Type 3 copper centers of laccase, which inhibits the binding of O<sub>2</sub>, the terminal electron acceptor, and therefore inhibits the activity at a higher pH because of the increased amount of OH<sup>-</sup> ions [59].

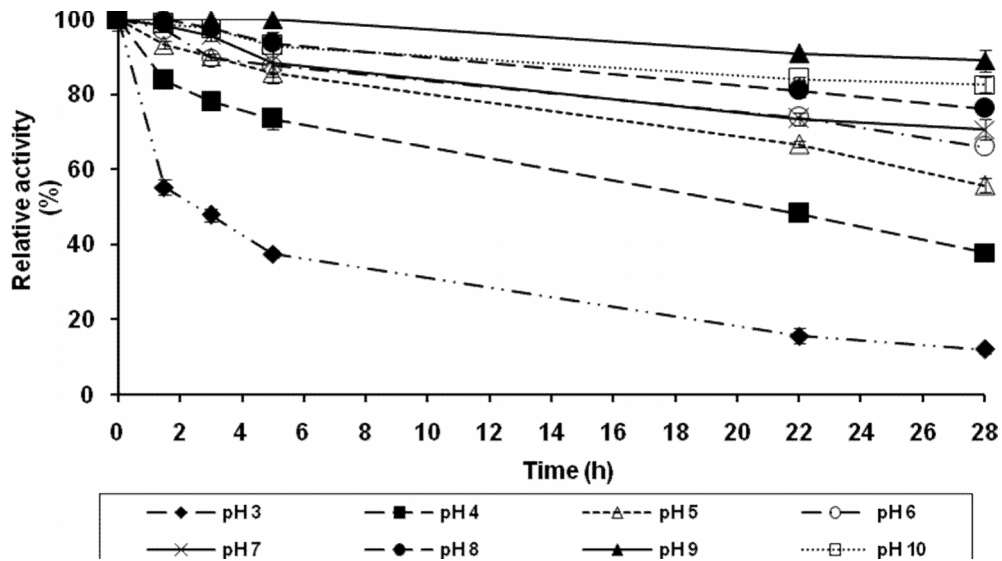


**Fig. 4. Effect of pH value on the activity of the *P. ostreatus* ARC280 purified laccase**  
 Reaction mixture contained: substrate (SGZ, 0.1  $\mu$ mole; DMP, 0.5  $\mu$ mole or ABTS, 0.5  $\mu$ mole); purified enzyme, 1.0  $\mu$ g protein; citrate-phosphate buffer pH (as indicated), 90  $\mu$ moles; total volume, 2ml. Values are means of three replications  $\pm$  standard deviation

The results obtained, indicated that the purified laccase from *P. ostreatus* ARC280 is stable at different pH values and more stable at alkaline range and it retains about 37.42, 73.51, 85.65, 87.70, 88.49, 93.65, 92.86 and 100.00 % of the initial activity after 5 h of incubation at pH 3.0, 4.0, 5.0, 6.0, 7.0, 8.0, 9.0 and 10.0, on the other hand, the enzyme retains about 11.97, 37.84, 55.74, 65.87, 70.63, 76.19, 88.98 and 82.54 % after incubation for 28h at the same pH values, respectively (Fig. 5). In contrast, Guo et al. [51] found that, LacI, LacII, and LacIII isozymes from the white rot fungus *Trametes* sp. HS-03 had considerable stability behavior at pHs of 2.5 to 4.5, 3.5 to 5.5 and 2.0 to 4.5, respectively, after incubation for 4 h, and at pH values higher than 7.0, all the laccase isozymes lost their activity very rapidly. Luna et al. [53], stated that the laccase from *P. ostreatus* showed greater stability at pH 7, with value of residual activity of 42.1% after 15 hours of incubation.

### 3.5 Effect of the Nature of Buffering System

Concerning the nature of the buffer systems, the results obtained indicated that the optimum laccase activity of *P. ostreatus* ARC280 was recorded with citrate-phosphate buffer system (406.67U/ml), followed by citrate buffer (331.67U/ml), succinate buffer (325.00U/ml) and maleate buffer (260.00U/ml). The lowest laccase activity could be detected with phthalate-NaOH buffering system (228.33U/ml).



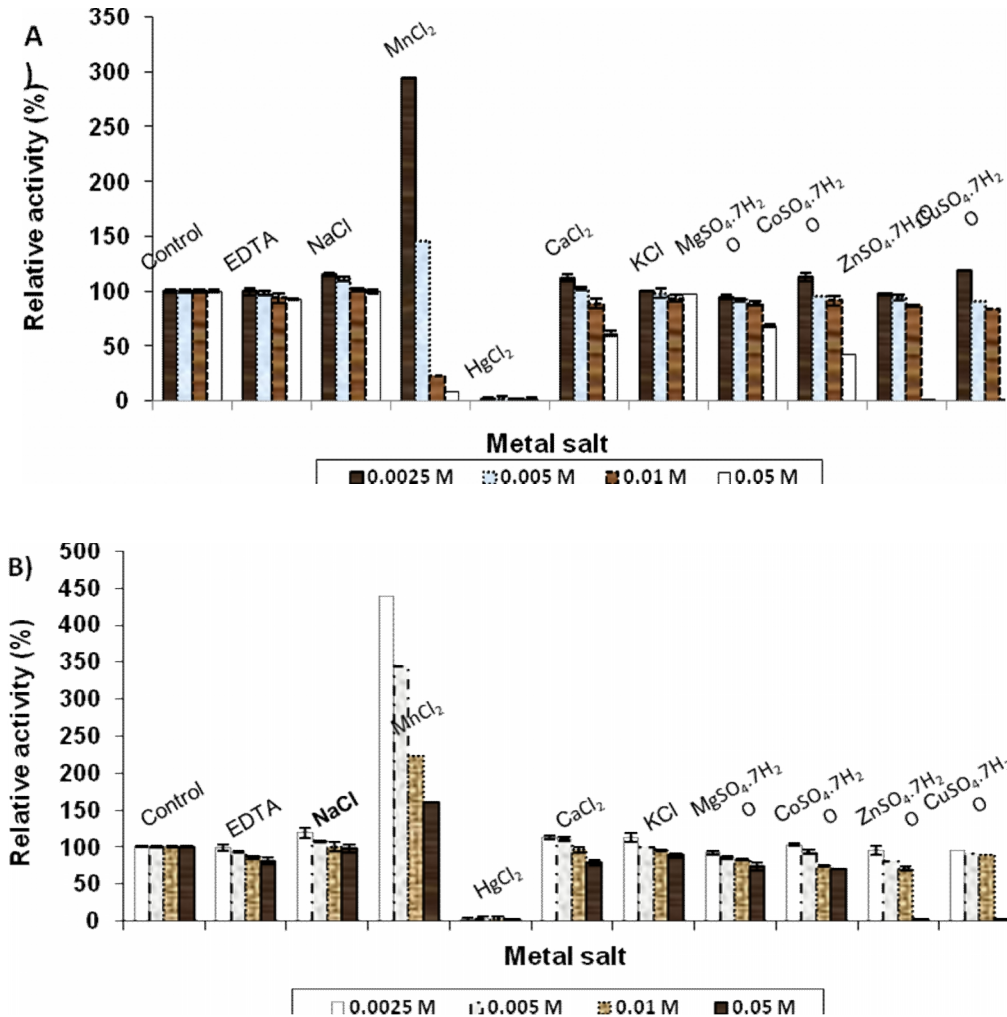
**Fig. 5. pH stability of the *P. ostreatus* ARC280 purified laccase**

Reaction mixture contained: SGZ, 0.1  $\mu$ mole; purified enzyme, 1.0 $\mu$ g protein; buffers (citrate phosphate buffer (pH 3.0 - 7.0), phosphate buffer (pH 8.0) and bi-carbonate buffer (pH 9.0 - 10.0)), 90  $\mu$ moles; incubation time of the enzyme in the absence of substrate, as indicated; total volume, 2 ml. Values are means of three replications  $\pm$  standard deviation

### 3.6 Effect of EDTA and Different Metal Salts

Chelating agent EDTA inhibited the enzyme only at higher concentrations which was in accordance with the earlier studies by other researchers [37]. Results obtained in this study indicated that the addition of EDTA at concentrations of  $2.5 \times 10^{-3}$ ,  $5 \times 10^{-3}$ ,  $1 \times 10^{-2}$  and  $5 \times 10^{-2}$  M caused slight inhibition to laccase activity according to the strength of EDTA concentrations used to give residual activities of 100.00, 98.42, 94.21 and 92.63%, respectively after 1h of incubation (Fig. 6a) and 99.32, 93.62, 86.32 and 81.05%, respectively after 24h (Fig. 6b). In agreement with our results, Zapata-Castillo et al. [50] found that the activity of *Trametes hirsuta* Bm-2 laccase is partially resistant to EDTA (10 mM). On the other hand, Asgher et al. [44] reported that EDTA inhibited the purified laccase enzyme from *Trametes versicolor* IBL-04. Pozdnyakova et al. [46], stated that EDTA, a weak inhibitor of fungal laccases, reduced the activity of the laccase from *P. ostreatus* D1 enzyme by 60% at a concentration of 100mM. Few exceptions such as laccases of *Marasmius quercophilus* [60] and *Phellinus ribis* [61] have been described, which are only inhibited by high concentrations of EDTA.

The obtained results indicated that, the addition of  $ZnSO_4 \cdot 7H_2O$  and  $CuSO_4 \cdot 5H_2O$  at a concentration of  $5 \times 10^{-2}$  M caused complete inhibition of laccase activity, while as  $HgCl_2$  caused complete inhibition at all tested concentrations. On the other hand, when  $MnCl_2 \cdot 4H_2O$  was added at concentrations of  $5 \times 10^{-3}$  and  $2.5 \times 10^{-3}$  M caused laccase activation by about 50 and 190 % increase, respectively (Fig. 6a). The prolonged exposure (24h) of the purified enzyme to metal salts indicated that laccase is more stable at the lowest metal ions concentration ( $2.5 \times 10^{-3}$  M). It gives 330% activation when incubated with  $Mn^{+2}$  at the previously mentioned concentration (Fig.6b).

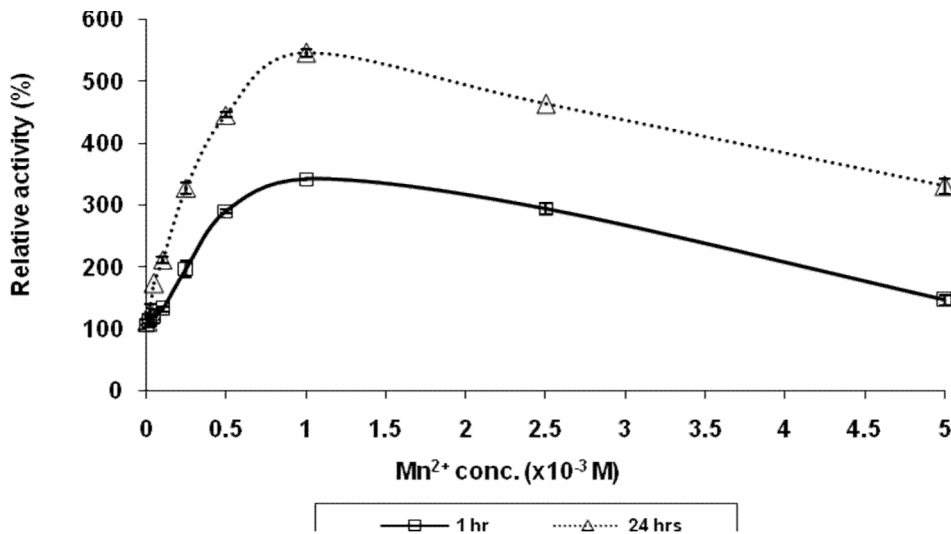


**Fig. 6. Effect of EDTA and different metal salts on the *P. ostreatus* ARC280 purified laccase. A) After 1 h and B) after 24 h of incubation**

Reaction mixture contained: SGZ, 0.1  $\mu$ mole; purified enzyme, 1.0 $\mu$ g protein; EDTA or metal salt, (as indicated) and citrate phosphate buffer pH 6.0, 90  $\mu$ moles; total volume, 2ml. Values are means of three replications  $\pm$  standard deviation

The activation effect caused by copper ions on laccase activity was mentioned with several reports on the metal ion requirements of laccases [40,62]. The positive response to copper is because laccases contain at least two types of copper centres; the first is a mononuclear centre that serves as the site for substrate oxidation and the second is a trinuclear centre where the reduction of oxygen to water occurs [63]. Despite this fact, not all laccases are stimulated by copper ions. For instance, laccase from *P. ostreatus* strain 10969 was reported to be inhibited by copper ions [64] in agreement with the present results. The observed ability of manganese ions to activate *P. ostreatus* ARC280 laccase is in general agreement with the findings on laccases from other sources [40,62].

According to Lundell and Hatakka [65], manganese ions participate either as reducing or oxidizing agents in laccase-catalyzed reactions. In the present investigation,  $Mn^{2+}$  caused an increase in the enzyme activity. The potent effect of different  $Mn^{2+}$  concentrations range ( $1.0 \times 10^{-5}M - 5.0 \times 10^{-3}M$ ) were tested. Laccase activation was increased by increasing  $Mn^{2+}$  concentration up to  $1.0 \times 10^{-3}M$  which gave about 240 and 445% activity increase after incubation for 1 and 24h, respectively (Fig. 7). Laccase activity then decreased by increasing  $Mn^{2+}$  concentration above this value. In agreement with current results, Adamafo et al. [40] reported the stimulatory effect of manganese on *P. ostreatus* strain EM-1 laccase contrasts sharply with the reported inhibition of *Daedalea quercina* laccase by  $Mn^{2+}$  [66].



**Fig. 7. Effect of different concentrations of  $Mn^{2+}$  on the *P. ostreatus* ARC280 purified laccase**

Reaction mixture contained: SGZ, 0.1  $\mu$ mole; purified enzyme, 1.0  $\mu$ g protein;  $Mn^{2+}$ , (as indicated) and citrate phosphate buffer pH 6.0, 90  $\mu$ moles; total volume, 2ml. Values are means of three replications  $\pm$  standard deviation

### 3.7 Substrate Specificity and Kinetic Constants

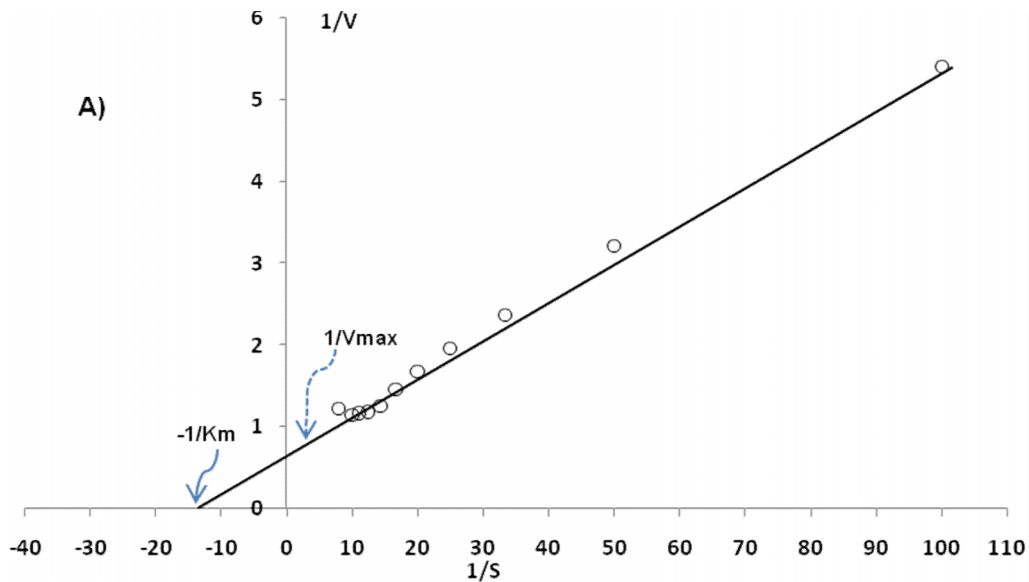
The reaction rate and the substrate affinity greatly varied depending on the nature of substrate. There is a difference in terms of reactivity of the enzyme towards different compounds. Tinoco et al. [47] stated that ABTS and SGZ were shown to be good substrates for laccases from all the *P. ostreatus* strains, and the apparent affinity constants ( $K_m$ ) also showed significant differences between the different strains, from 8 to 80mmol  $l^{-1}$  for ABTS, from 12 to 52 $\mu$ mol  $l^{-1}$  for SGZ and from 0.46 to 6.61mmol  $l^{-1}$  for guaiacol. In this connection, Patel et al. [37] found that the  $K_m$  values for *P. ostreatus* HP-1 laccase with ABTS and DMP were 46.51 and 400mM, respectively. The laccase activity against SGZ was much higher than that against ABTS and DMP, respectively. Results also revealed that SGZ was the best substrate for the purified laccase of *P. ostreatus* ARC280 (Table 2).

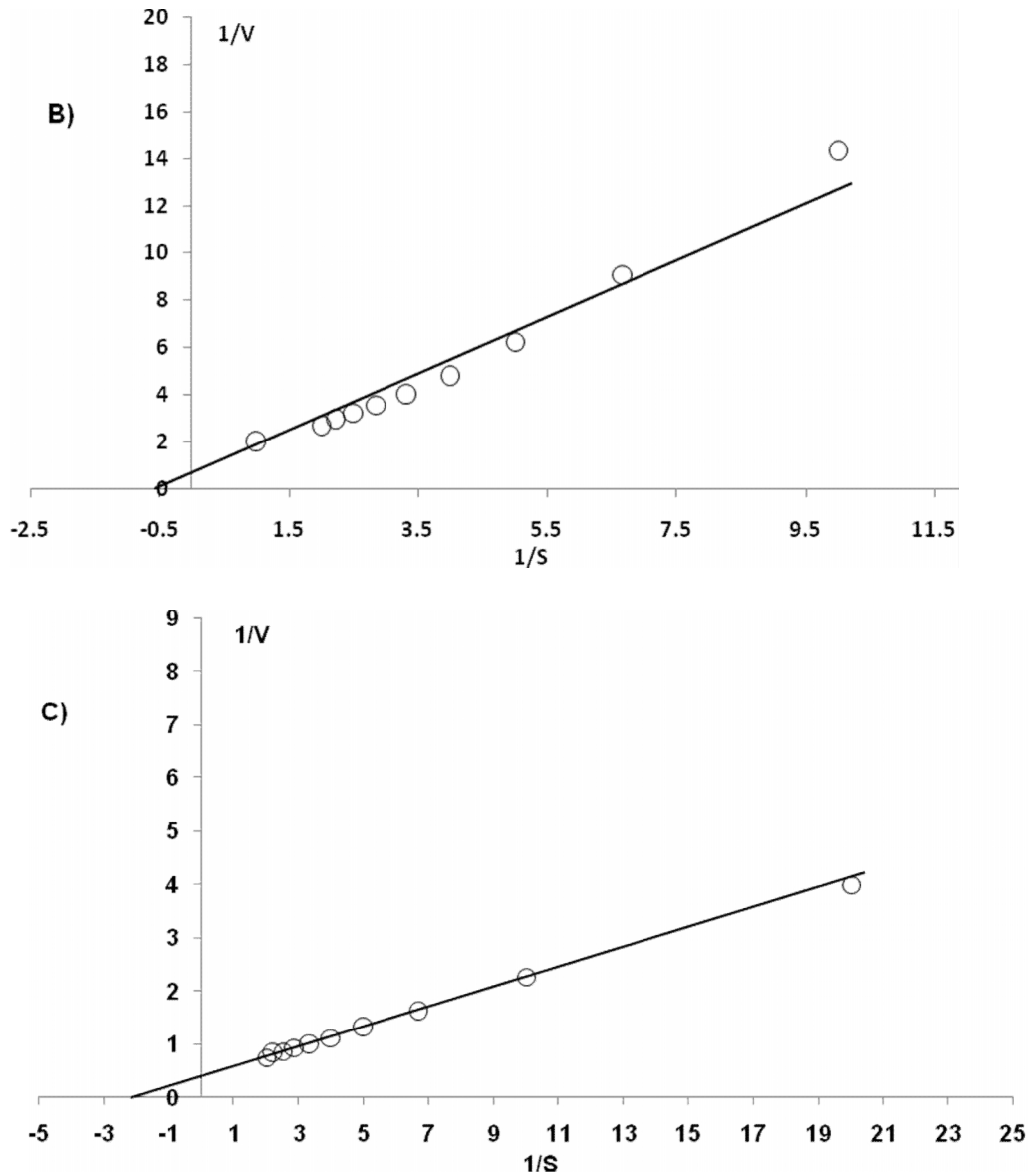
**Table 2. Substrate specificity and kinetic constants for the oxidation of various substrates by the *P. ostreatus* ARC280 purified laccase**

Substrate	Optimal pH	Extinction coefficient ( $M^{-1} cm^{-1}$ )	Relative activity (%)	$K_m$ ( $\mu M$ )	$V_{max}$ ( $\mu mol min^{-1}$ )	$K_m/V_{max}$
SGZ	6.0	65000	100.00	0.074	1.563	0.047
DMP	4.5	49600	8.05	2.857	2.500	1.143
ABTS	3.0	36000	50.57	0.476	2.632	0.181

Reaction mixture (2ml) contained: substrate, 0.1  $\mu mole$  SGZ, 0.5  $\mu mole$  DMP or 0.5  $\mu mole$  ABTS; purified enzyme, 1.0  $\mu g$  protein and citrate phosphate buffer pH (as indicated), 90  $\mu moles$

$K_m$  and  $V_{max}$  were determined using various concentrations of different substrates (SGZ, DMP and ABTS) with the purified *P. ostreatus* ARC280 laccase (Figs. 8a, b and c).  $K_m$  and  $V_{max}$  were calculated and found to be 0.074, 2.857 and 0.476  $\mu M$  and 1.563, 2.500 and 2.632  $\mu mol min^{-1}$  for SGZ, DMP and ABTS, respectively (Table 2). In agreement with our results, Stoilova et al. [67] found that laccase from *Trametes versicolor* has  $K_m$  value for SGZ substrate lower than the value of ABTS substrate and stated that, the  $K_m$  index for SGZ substrate is lower than the value of the index established for ABTS substrate, pointing out the greater affinity of the investigated laccase to SGZ. In contrast, Baldrian [55] stated that in general, laccases are known to possess very wide range substrate affinities; however the most of the laccases characterized show greater affinity for ABTS. In this connection, Asgher et al. [44] reported that on using ABTS as a substrate, the enzyme showed  $V_{max}$  of 780 Units/ml with a corresponding  $K_m$  value of 73  $\mu M$ . Zapata-Castillo et al. [50] found that the values of apparent  $K_m$  are 68 and 164  $\mu M$  and  $V_{max}$  values are 14 and 4.65 Units/ml for ABTS and DMP, respectively for *Trametes hirsuta* Bm-2 laccase.





**Fig. 8. Lineweaver–Burk plot of the reciprocal of initial velocities and different laccase substrates concentrations**

A) SGZ: Syringaldazine (4-hydroxy-3,5- dimethoxy-benzaldehyde azine), B) DMP: 2, 6-dimethoxyphenol and C) ABTS: 2, 2'-azino-bis-3-ethylbenzthiazoline-6-sulfonic acid

### 3.8 Anti-Proliferative (Anticancer) Activity

A few of laccases were reported to manifest some bioactivities such as anti-proliferative activity toward tumor cells [68]. Potential successes in cancer treatment using bioactive metabolites isolated from medicinal mushrooms have shown as biological immunotherapeutic agents stimulating the immune system against cancer cells. These

bioactive metabolites also act as an effective source of anticancer agents, capable of interfering with cellular signal transduction pathways linked to cancer development and progression [69].

In the present study, the in-vitro bioassay cytotoxic effect of *P. ostreatus* ARC280 laccase crude enzyme on the growth of four human tumor cell lines namely HePG2, HCT116, A549 and MCF7 revealed that, the crude laccase enzyme extract of *P. ostreatus* ARC280 at a concentration of 100 µg/ml, has anti proliferative activity against HePG2 and MCF7 cell lines growth in-vitro with percentage reduction of 33.4 and 45.9%, respectively. In contrast, the *P. ostreatus* ARC280 crude laccase enzyme extract have no anti proliferative activity against HCT116 and A549 cell lines growth in-vitro (Table 3).

**Table 3. In-vitro bioassay of *P. ostreatus* ARC280 laccase on human tumor cell lines**

Sample	Anticancer activity (%) at conc. of 100µg/ml			
	HePG2 <sup>a</sup>	HCT116 <sup>b</sup>	A549 <sup>c</sup>	MCF7 <sup>d</sup>
Crude laccase	33.4	0	0	45.9
Pure laccase	16.8	23.4	15.2	23.4
Negative control	0	0	0	0

<sup>a</sup> Human hepatocellular carcinoma cell line

<sup>b</sup> Colon cell line

<sup>c</sup> Lung carcinoma cell line

<sup>d</sup> Human caucasian breast adenocarcinoma

On the other hand, the treatment of the four human tumor cell lines previously mentioned with *P. ostreatus* ARC280 purified laccase (100µg/ml) resulted in the reduction of different tested cell lines growth in-vitro with a percentage reduction of 16.8, 23.4, 15.2 and 23.4 % for HePG2, HCT116, A549 and MCF7, respectively (Table 3). The highest antitumor activity was recorded with the *P. ostreatus* ARC280 crude laccase enzyme extract toward MCF7 cell line growth in-vitro (45.9%) when compared with the negative control. From the obtained results it could be concluded that, *P. ostreatus* ARC280 laccase can be used for developing the therapy of different types of tumors. In this connection, Hu et al. [70] reported that the purified laccase from an edible mushroom, *Agrocybe cylindracea* has highly potent antiproliferative activity against hepatoma HepG2 cells and breast cancer MCF-7 cells, where at the concentrations of 1.25, 2.5, 5.0, and 10µM, inhibited proliferation of HepG2 cells by 7.8, 30.2, 46.4, and 78.5%, respectively, and MCF7 cells by 7.2, 22.7, 41.3, and 70.6%, respectively.

### 3.9 Anti-Microbial Activity

Beyond the chemical compounds that can be used as antimicrobial agents, enzymes are also interesting as antimicrobial agents. Furthermore, because of their proteinaceous nature, enzymes are considered to be environmentally safe [71]. In this connection, Ibrahim et al. [72] stated that, the effect of enzymes as antimicrobial agents is mainly due to the electrochemical mode of action to penetrate cell wall of microorganisms, thereby causing leakage of essential metabolites and physically disrupting other key cell functions.



Table 4. Anti-microbial activity of *P. ostreatus* ARC280 laccase

Sample	Diameter of inhibition zone (mm)		
	<i>B. mycooides</i>	<i>E. coli</i>	<i>C. albicans</i>
Crude laccase	43	30	27
Partial purified laccase ((NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> precipitation)	17	23	25
Purified laccase	33	35	37

Table 5. A comparison of purified laccase properties from *P. ostreatus* ARC280 with other reported *P. ostreatus* strains

Source	M.w. (kDa)	Temperature (°C)	Thermal stability (Residual activity)	pH stability (Residual activity)	pH optimum	K <sub>m</sub>		References	
<i>P. ostreatus</i> ARC280	85 kDa	50	96, 80, 60% after 120 min at 40, 50, 60°C	88.49, 93.65, 92.86 and 100.00 % after 5 h at pH 7.0, 8.0, 9.0 and 10.0	ABTS	3.0	ABTS	0.476 µM	The present study
					DMP	4.5	DMP	2.857 µM	
					SGZ	6.0	SGZ	0.074 µM	
<i>P. ostreatus</i> HP-1	68	50	50 % after 15 min at 60°C	ND	ABTS	4.5	ABTS	46.51 mM	Patel et al. [37]
					DMP	3.5	DMP	400mM	
					Guaiacol	5.5	Guaiacol	100mM	
					O-dianisidine	3.5	O-dianisidine	23.52 mM	
<i>P. ostreatus</i> D1	64	ND	ND	ND	ABTS	4.0	ABTS	0.11 mM	Pozdnyakova et al. [46]
					DMP	4.0	DMP	0.43 mM	
					SGZ	7.0	SGZ	0.0087 mM	
					Pyrocatechol	8.0	Pyrocatechol	3.65 mM	
<i>P. ostreatus</i> 10969	40	50	10% after 10 min at 60°C	10% after 60 min at pH 7.0	ABTS	4.0	ABTS	0.31 mmol / l	Liu et al. [74]
<i>P. ostreatus</i> strain EM-1	78	50	22.6 % after 60 min at 50°C	ND	DMP	5.0	ND		Adamafo et al. [40]
<i>P. ostreatus</i> lacc POXA1	61	45–65	t <sub>1/2</sub> = 200 min at 60°C, pH 7.0	t <sub>1/2</sub> = 24 h at pH 3.0	ABTS	3.0	ABTS	0.09 mM	Palmieri et al. [38, 48]
					DMP	3-5	DMP	2.1 mM	
					SGZ	6.0	SGZ	1.3 mM	
<i>P. ostreatus</i> lacc POXC		50 – 60	t <sub>1/2</sub> = 30 min at 60°C, pH 7.0	t <sub>1/2</sub> = 30 min at pH 3.0	ABTS	3.0	ABTS	2.8 mM	Palmieri et al. [38, 48]
					Guaiacol	6.0	Guaiacol	1.2 mM	
					DMP	3-5	DMP	2.3 mM	
					SGZ	6.0	SGZ	2.0 mM	
<i>P. ostreatus</i> lacc POXA2	67	25–35	t <sub>1/2</sub> = 10 min at 60°C, pH 7.0	t <sub>1/2</sub> = 2 h at pH 3.0	ABTS	3.0	ABTS	1.2 mM	Palmieri et al. [38, 48]
					Guaiacol	6.0	Guaiacol	3.1 mM	
					DMP	6.5	DMP	7.4 mM	
					SGZ	6.0	SGZ	1.4 mM	
<i>P. ostreatus</i>	ND	ND	27.5 % at 39°C after 15 h	42.1 % at pH 7 after 15 h	ND		ND	Luna et al. [53]	
<i>P. ostreatus</i> strains	ND	ND	ND	ND	ND		ABTS	8 - 80 mmol / l	Tinoco et al. [47]
							SGZ	12 - 52 µmol / l	
							Guaiacol	0.46 - 6.61 mmol / l	

In the present study, the results of antimicrobial activity indicated that the pure laccase produced by *P. ostreatus* ARC280 gave the highest Gram-negative antibacterial and antifungal activities against *E. coli* (35 mm) and *C. albicans* (37 mm) respectively, expressed as the width of inhibition zone. Optimum Gram positive antibacterial activity against *B. mycooides* (43 mm) was found with the crude preparation of *P. ostreatus* ARC280 laccase (Table 4) above. In agreement with the present results, Arul Diana Christie and Shanmugam [73] reported that the laccase enzyme could inhibit both Gram positive and Gram negative bacteria.

Table (5) above summarizes the properties of purified laccase from *P. ostreatus* ARC280 and other reported *P. ostreatus* strains. The table indicates the stability behavior of *P. ostreatus* ARC280 laccase under study compared with some of other reported *P. ostreatus* laccases.

#### 4. CONCLUSION

In view of the results obtained, it can be concluded that *P. ostreatus* ARC280 laccase seems to be a prospective enzyme for further biotechnological exploitation such as anticancer and antimicrobial activity applications.

#### COMPETING INTERESTS

Authors have declared that no competing interests exist.

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